## SYNTHESES OF " THIIRANORADIALENES "

Wataru Ando\*, Yukio Haniu and Toshikazu Takata

Department of Chemistry, The University of Tsukuba
Sakura-mura, Ibaraki 305, Japan

Summary: A stable tetramethylthiiranoradialene, 2,3-diisopropylidenethiirane  $\underline{6}$ , was prepared by pyrolysis of lithium salt of tosyl hydrazone of 2,2-dimethyl-4-isopropylidene-3-thietanone. The corrsponding sulfoxide  $\underline{7}$  was also prepared by oxidation of 6.

Special interest has long been focused on the syntheses of [3]  $\sim$  [6] radialenes. Nature of these unique  $\pi$ -systems has been associated with the aromaticity as well as unusual reactivity. Although recently preparation of "furanoradialene" has been reported no heteroradialene has been prepared for three-membered ring or any closely related derivatives. We present here first preparations of three-membered sulfur-radialene, "thiiranoradialene" 6, and its S-oxide 7, in connection with our carbene chemistry.

2,2-Dimethyl-4-isopropylidene-3-thietanone  $\underline{1}$ , prepared by Kellogg et al., was treated with excess tosyl hydrazine for a few minutes in the presence of catalytic amount of BF $_3$  etherate in EtOH at 50  $\sim$  70°C, yielding hydrazone  $\underline{2}$ . The purified hydrazone  $\underline{2}$  was treated with equimolar of BuLi at -70°C in THF under dry conditions. Removal of the solvent afforded lithium salt of the hydrazone  $\underline{2}$  which then was subjected to the usual bulk vacuum pyrolysis at  $130 \sim 160$ °C (at ca  $10^{-4}$  torr). During the pyrolysis, nitrogen was

NNHTs
$$\begin{array}{c}
 & \text{NNHTs} \\
 & \text{NNLiTs} \\
 & \text{S} \\
 &$$

evolved and colorless crystalline solid was collected in liq. nitrogen trap. The solid purified by silica gel column chromatography ( 60  $\sim$  70% yield ) showed m.p. of 42.5  $\sim$  43.5°C and is stable under the conditions. The spectral data of the solid: IR( cm  $^{-1}$ , CDCl  $_3$ ), 2960  $_{\rm w}$ , 2900  $_{\rm m}$ , 2850  $_{\rm w}$ , 1440  $_{\rm m}$ , 1360  $_{\rm w}$ , 1060  $_{\rm w}$ ;  $^{1}$ H-NMR(  $\delta$ , CDCl  $_3$ ), 2.05( s, 6H ), 1.97( s, 6H );  $^{13}$ C-NMR(  $\delta$ , CDCl  $_3$ ), 115.5, 111.3, 23.0, 22.7; MS( 70eV ), m/e 140( M  $^{+}$  ), and elemental analysis ( found: C: 68.69; H: 8.66; S: 22.59, required for C  $_8$ H  $_12$ S ) established unequivocally the structure of  $_6$ . The ring contraction of carbene (  $_4$  ) derived from the pyrolysis of lithium salt 3 may afford  $_6$  in good yield via intermediary bicyclic sulfur ylide,  $_5$ .

Oxidation of thiiranoradialene  $\underline{6}$  with equimolar of MCPBA in CH<sub>2</sub>Cl<sub>2</sub> at ca 0°C gave readily "thiiranoradialene S-oxide " $\underline{7}$  in quantitative yield. The sulfoxide  $\underline{7}$  was colorless crystals having m.p.  $58.0 \sim 58.5$ °C and more labile than sulfide  $\underline{6}$ , slowly decomposing to unidentified substances even at -20°C. The structure of  $\underline{7}$  was confirmed by following spectral data: IR( cm<sup>-1</sup>, CDCl<sub>3</sub>)  $1035(v_{S-O})$ ;  $^1_{H-NMR}(\delta, CDCl_3)$ , 2.22(s, 6H), 2.17(s, 6H);  $^{13}_{C-NMR}(\delta, CDCl_3)$ , 131.2, 125.2, 23.4, 23.0.

## REFERENCES AND FOOTNOTE

- 1) Barkorich, A. J., Strauss, E. S., and Vollhart, K. P. C., J. Am. Chem. Soc., 1977, 99, 8321, and references cited therein.
- Jullien, J., Pechine, J. M., Perez, F. and Piade, J. J., Tetrahedron Lett., 1980, 21, 611.
- 3) (a) Ando, W., Acc. Chem. Res., 1977, 10, 179. (b) Ando, W., "The Chemistry of Diazonium and Diazo-compounds ", ed. by Patai, S., part 1, 1978, P. 341
- 4) Bolster, J. and Kellogg, R. M., J. Org. Chem., 1980, 45, 4804.
- 5) Purified yield of  $\underline{2}$  by column chromatography (silica gel,  $\text{CH}_2\text{Cl}_2$ ) was ca 15% and ca 30% of 1 was recovered.
- 6) Hortmann, A. G. and Bhattacharjya, A., J. Am. Chem. Soc., 1976, 98, 7081.

(Received in Japan 23 September 1981)